- PRINCE, E., WLODAWER, A. & SANTORO, A. (1978). J. Appl. Cryst. 11, 173–178.
- RICHARDS, F. M. & WYCKOFF, H. W. (1971). *The Enzymes*, Vol. 4, edited by P. D. BOYER, pp. 647–806. New York: Academic Press.
- SANTORO, A. & WLODAWER, A. (1980). Acta Cryst. A36, 442–450.
- SCHOENBORN, B. P. (1969). Nature (London), 224, 143-146.
- SCHOENBORN, B. P. & NUNES, A. C. (1972). Annu. Rev. Biophys. Bioeng. 1, 529–552.
- SCHOENBORN, B. P., NUNES, A. C. & NATHANS, R. (1970). Ber. Bunsenges. Phys. Chem. 74, 1202–1207.
- SCHREIER, A. A. & BALDWIN, R. L. (1976). J. Mol. Biol. 105, 409–426.
- TORII, K., URATA, Y., IITAKA, Y., SAWADA, F. & MITSUI, Y. (1978). J. Biochem. (Tokyo), 83, 1239–1247.
- WODAK, S. Y., LIU, M. Y. & WYCKOFF, H. W. (1977). J. Mol. Biol. 116, 855–875.
- WYCKOFF, H. W., TSERNOGLOU, D., HANSON, A. W., KNOX, J. R., LEE, B. & RICHARDS, F. M. (1970). J. Biol. Chem. 245, 305–328.

Acta Cryst. (1980). B36, 1831–1835

Crystal Structures of 2,3,4a β ,6,7,8a β -Hexamethyl-4a,5,8,8a-tetrahydro-1-naphthoquin-4 α (and 4 β)-ol*

BY TREVOR J. GREENHOUGH AND JAMES TROTTER

Department of Chemistry, University of British Columbia, Vancouver, BC, Canada V6T 1W5

(Received 5 October 1979; accepted 10 March 1980)

Abstract

Crystals of the 4α -ol are triclinic, PI, a = 7.660 (1), b = 8.194 (1), c = 12.981 (2) Å, $\alpha = 75.17$ (1), $\beta = 79.57$ (1), $\gamma = 65.19$ (1)°, Z = 2, R = 0.043 for 2438 reflexions. The 4β -ol is monoclinic, $P2_1/n$, a = 7.671 (1), b = 16.877 (2), c = 10.994 (1) Å, $\beta = 92.27$ (2)°, Z = 4, R = 0.041 for 1640 reflexions. The molecules have different, ring-flipped conformations, so that in each molecule the hydroxyl group is pseudo-equatorial to the cyclohexenone ring. The two conformations produce differing intramolecular nonbonded contacts, which account for the varying photochemical reactivities in the solid state, and suggest that the photorearrangements in solution proceed from different conformations from those observed in the crystal lattice. Both structures contain intermolecular $O(4)-H\cdots O(1)$ hydrogen bonds.

Introduction

A study of the photochemistry of various substituted $4a\beta$,5,8,8 $a\beta$ -tetrahydro-1-naphthoquin-4-ol systems [*e.g.* (I), (II)] has revealed reactivity differences, with either no solid-state reaction or formation of different photoproducts in solution and in the solid state (Appel,

Greenhough, Scheffer, Trotter & Walsh, 1980). To establish the reasons for these different results and to provide structural data which might indicate the factors influencing the reaction pathways, a structural study of the starting substrates has been undertaken. The present paper describes the structures of 2,3,4 $\alpha\beta$,6,7,8 $\alpha\beta$ -hexamethyl-4 α ,5,8,8 α -tetrahydro-1naphthoquin-4 α -ol (I) and -4 β -ol (II).



Experimental

Recrystallization from petroleum ether/ethanol gave well formed colourless chunks of the 4α -ol and large colourless plates of the 4β -ol; fragments were cut, dimensions $0.3 \times 0.3 \times 0.3$ of са mm and $0.6 \times 0.2 \times 0.2$ mm, respectively. Preliminary unit-cell data were obtained from precession photographs, and accurate unit-cell parameters (Table 1) were determined by a least-squares analysis of the setting angles of 25 reflexions automatically located and centred on an Enraf-Nonius CAD-4 diffractometer ($\theta = 12-21^{\circ}$ for the 4α -ol, $8-12^{\circ}$ for the 4β -ol; graphite-monochromatized Mo $K\alpha$ radiation). Space groups were

© 1980 International Union of Crystallography

^{*} IUPAC names: 4α (and 4β)-hydroxy-2,3,4a β ,6,7,8a β -hexamethyl-4a,5,8,8a-tetrahydro-1(4H)-naphthalenone.

	4α -ol	4β-ol
Formula	C ₁₆ H ₂₄ O ₂	C ₁₆ H ₂₄ O ₂
Crystal system	Triclinic	Monoclinic
Space group	$P\hat{I}(C_i^1)$	$P2_{1}/n (C_{2h}^{5})$
a (Å)	7.660(1)	7.671 (1)
b (Å)	8.194 (1)	16.877 (2)
c (Å)	12.981 (2)	10-994 (1)
α (°)	75-17 (1)	
β(°)	79-57 (1)	92.27 (2)
γ(°)	65.19(1)	-
Z	2	4
D_{x} (Mg m ⁻³)	1.16	1.16
Radiation	Mo Ka	Μο <i>Κ</i> α
Monochromator	Graphite	Graphite
Scan	$\omega - \theta$	ω-θ
Δω (°)	$1.4 + 0.35 \tan \theta$	$0.7 + 0.35 \tan \theta$
Aperture height (mm)	4	4
Aperture width (mm)	$1.5 + 1.0 \tan \theta$	$1.5 + 1.0 \tan \theta$
Maximum scan speed (deg min ⁻¹)	10.06	10.06
Prescan acceptance limit	30σ	30σ
Maximum counting time (s)	90	75
Maximum θ (°)	27.5	25
Number of check reflexions	3	3
Variation of checks (%)	±0.5	±0·7
Total reflexions	3254	2484
Reflexions $>3\sigma^*$	2438 (75%)	1640 (66%)
R	0.043	0.041
R _w	0.063	0.059
Goodness of fit	1.23	1.38

Table 1. Experimental data

* $\sigma^2(I) = S + B + (0.04S)^2$, S = scan count, B = background count.

determined from systematically absent reflexions, intensity statistics, and structure analyses.

The intensities were measured on the CAD-4 diffractometer; the experimental conditions are summarized in Table 1.

All non-hydrogen atoms were located by direct methods with use of MULTAN (Main, Hull, Lessinger, Germain, Declercq & Woolfson, 1978), with 496 and 493 |E| values > 1.55 and 1.45 for the 4 α -ol and 4β -ol, respectively, derived from a K-curve method (Ladd, 1978). Following anisotropic full-matrix leastsquares refinements of C and O, the H atoms were located from difference maps and were assigned isotropic thermal parameters. Scattering factors for C and O were taken from Cromer & Mann (1968) and those for H from Stewart, Davidson & Simpson (1965). The final refinements included all atomic parameters, and the function minimized was $\sum w(F_o - F_c)^2$, $w = 1/\sigma^2(F)$; for the 4 β -ol, an isotropic extinction parameter was refined, final value g = 1.9 (6) × 10⁴ (Becker & Coppens, 1974, 1975). Final values of $R = \sum |\Delta F| / \sum |F_o|$ and $R_w = \left[\sum w(F_o - F_c)^2 / \sum wF_o^2\right]^{\frac{1}{2}}$ for the reflexions >3 σ are given in Table 1. The mean and maximum shift/error on the final least-squares cycles were 0.01 and 0.23, respectively, for the 4α -ol, and 0.06 and 0.74 for the 4β -ol; weighting analyses confirmed the suitability of the chosen weights. Difference Fourier syntheses after the final cycles showed random fluctuations of up to ± 0.25 eÅ⁻³. Final positional parameters are listed in Table 2.*

Discussion

The tetrahydro-1-naphthoquin-4-ol ring system can exist in two low-energy conformations (A and B in Fig. 1); both have fused six-membered rings with half-chair conformations, and are folded and twisted about the C(4a)-C(8a) bond. They can be interconverted by ring 'flipping' involving single-bond rotations via a higherenergy conformer (C in Fig. 1), which has eclipsed bridgehead substituents and parallel double bonds. The flipping converts a pseudo-equatorial substituent at C(4) in the cyclohexenone ring of one conformer. These twist conformations are similar to that found in various substituted tetrahydro-1,4-naphthoquinones (Phillips & Trotter, 1977).

The molecular structures of the present hexamethyl-4 α - and -4 β -ol derivatives (Fig. 2) show that the two molecules have different ring-flipped conformations, Afor the 4 α -ol and B for the 4 β -ol [to illustrate the similarity of the fused-ring systems in the two conformations, the molecules shown in Fig. 2 are enantiomorphic at the bridgehead C(4a) and C(8a)

^{*} Lists of structure factors, thermal parameters, bond distances involving hydrogen atoms, torsion angles, and packing diagrams for both compounds have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 35181 (59 pp.). Copes may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.



Fig. 1. Conformations of tetrahydro-1-naphthoquin-4-ols. Observed conformations in the crystal lattice are A for the 4 α -ol with R = OH, and B for the 4 β -ol with R' = OH.

Table 2. Final positional parameters (fr	ractional $ imes 10^5$, for H $ imes$	(10 ⁴) with e.s.d.'s in parentheses
--	--	---

	4a-ol (PĪ)			4β -ol ($P2_1/n$)			
	x	У	Ζ	x	у	Ζ	
C(1)	84844 (16)	79109 (17)	34940 (10)	23525 (25)	29745 (13)	50503 (20)	
C(2)	70973 (17)	75859 (17)	43917 (10)	30356 (26)	37019 (12)	56165 (19)	
C(3)	54095 (17)	76387 (16)	42059 (10)	47694 (26)	38106 (12)	57773 (19)	
C(4)	48850 (16)	78796 (17)	30899 (10)	60782 (25)	32220 (13)	53083 (22)	
C(4a)	66129 (16)	75174 (16)	22438 (9)	53351 (24)	26658 (12)	43166 (18	
C(5)	78548 (17)	54446 (17)	24141 (11)	66313 (30)	19854 (14)	41331 (24)	
C(6)	98250 (17)	49275 (18)	18236 (10)	67035 (26)	13571 (12)	51005 (22)	
C(7)	106345 (18)	61379 (20)	14180 (11)	54877 (27)	12909 (12)	59240 (21)	
C(8)	96365 (20)	81025 (20)	15396 (12)	39552 (28)	18451 (13)	59395 (22)	
C(8a)	78643 (17)	85251 (17)	23569 (10)	36062 (23)	23105 (12)	47483 (19)	
C(21)	76943 (27)	72481 (27)	54990 (12)	17010 (42)	42919 (19)	60185 (35)	
C(31)	38719 (25)	74983 (26)	50877 (13)	55233 (43)	45070 (17)	64572 (31)	
C(41)	58558 (24)	81759 (25)	11323 (12)	50404 (41)	31267 (20)	31214 (25)	
C(61)	107939 (24)	29334 (23)	17496 (17)	82408 (40)	7984 (20)	50483 (43)	
C(71)	125842 (27)	56841 (33)	7931 (19)	55028 (53)	6791 (21)	69200 (37)	
C(81)	67616 (24)	106291 (20)	22021 (15)	27599 (37)	17529 (19)	37990 (31)	
O(1)	100292 (14)	78187 (18)	36788 (9)	7758 (19)	28753 (11)	48730 (20)	
O(4)	39954 (14)	66683 (15)	30720 (9)	75064 (21)	36710 (10)	48953 (22)	
H(O4)	2806 (31)	7147 (27)	3220 (16)	8332 (44)	3380 (20)	4900 (28)	
H(4)	3921 (20)	9199 (20)	2898 (12)	6444 (29)	2914 (13)	6000 (21)	
H(51)	7157 (22)	4851 (22)	2168 (13)	6386 (31)	1744 (15)	3371 (24)	
H(52)	7962 (21)	4950 (22)	3183 (13)	7834 (33)	2221 (14)	4077 (22)	
H(81)	10548 (24)	8443 (22)	1726 (13)	2920 (32)	1561 (14)	6122 (20)	
H(82)	9267 (26)	8909 (26)	872 (15)	4146 (30)	2216 (15)	6685 (23)	
H(211)	9006 (40)	6842 (36)	5485 (22)	860 (46)	4036 (21)	6487 (32)	
H(212)	7390 (40)	6226 (42)	6042 (25)	2192 (46)	4777 (25)	6208 (35)	
H(213)	6947 (36)	8370 (37)	5813 (21)	977 (56)	4469 (26)	5387 (40)	
H(311)	3984 (44)	6351 (47)	5401 (26)	4765 (58)	4760 (26)	6908 (43)	
H(312)	3752 (46)	8134 (47)	5629 (27)	6499 (54)	4364 (23)	6972 (38)	
H(313)	2551 (43)	8215 (40)	4884 (23)	6077 (55)	4861 (27)	5953 (43)	
H(411)	6818 (28)	7927 (27)	585 (16)	4758 (38)	2757 (18)	2479 (29)	
H(412)	5054 (28)	9550 (30)	983 (15)	6058 (42)	3422 (18)	2918 (26)	
H(413)	5056 (34)	7607 (32)	1075 (18)	4138 (37)	3550 (17)	3204 (24)	
H(611)	12129 (35)	2595 (32)	1497 (18)	8468 (50)	732 (23)	4219 (38)	
H(612)	10154 (31)	2648 (29)	1290 (18)	7980 (50)	274 (26)	5362 (40)	
H(613)	10530 (32)	2174 (33)	2425 (20)	9326 (57)	1038 (27)	5490 (38)	
H(711)	12878 (37)	4741 (38)	425 (22)	6608 (56)	449 (26)	7039 (35)	
H(712)	12572 (38)	6745 (40)	272 (24)	4746 (63)	283 (28)	6736 (45)	
H(713)	13580 (38)	5285 (37)	1269 (22)	5160 (58)	909 (29)	7752 (45)	
H(811)	5618 (24)	11043 (23)	2705 (14)	3521 (41)	1272 (21)	3600 (30)	
H(812)	7697 (26)	11155 (26)	2279 (14)	1703 (43)	1540 (18)	4100 (26)	
H(813)	6421 (27)	11087 (28)	1510 (18)	2392 (53)	2063 (23)	3064 (38)	

atoms; the crystals are racemic]. The hydroxyl groups occupy the more sterically favoured pseudo-equatorial positions in both molecules, and this is probably the main feature which controls the conformations. The consequence of these conformations is that there is a close intramolecular non-bonded approach in the 4 α -ol of H(52) to the enone carbon atoms C(2) and C(3), with well separated, non-parallel C=C bonds; in the 4 β -ol the close approach is from H(82) to C(2) and C(3). The degree of twist in the molecules is indicated by the bridgehead torsion angles: 4α -ol, C(41)–C(4a)– C(8a)–C(81) = 59.9 (1), C(5)–C(4a)–C(8a)–C(1) = $-63.2 (1)^\circ$; 4β -ol, C(41)–C(4a)–C(8a)–C(81) = $58.6 (2), C(4)-C(4a)-C(8a)-C(8) = -60.0 (2)^\circ$. The twists are thus similar in the two molecules, and to those in the tetrahydronaphthoquinone analogue, 60.0(6) and $-61.4(5)^{\circ}$ (Phillips & Trotter, 1976).

In the 4α -ol the C(2)···H(52) and C(3)···H(52) distances are 2.78 (2) and 2.72 (2) Å, respectively, not significantly different from each other, but shorter than the normal van der Waals contact distance of 2.90 Å. In the 4β -ol the C(2)···H(82) and C(3)···H(82) distances are 2.88 (2) and 2.92 (2) Å, respectively, significantly longer than the enone C···H(52) distances in the 4α -ol, and comparable to the normal van der Waals distance. The solid-state photochemical reaction of the 4α -ol proceeds *via* initial H(52) abstraction by the β enone carbon, C(3), while the 4β -ol is photochemically unreactive in the solid state (Appel, Greenhough, Scheffer, Trotter & Walsh, 1980).



Fig. 2. Stereodiagrams of 2,3,4 $\alpha\beta$,6,7,8 $\alpha\beta$ -hexamethyl-4a,5,8,8atetrahydro-1-naphthoquin-4 α -ol (top) and -4 β -ol (bottom). Thermal ellipsoids are at the 50% probability level, and methyl H atoms are omitted for clarity.

The C(2)-C(3)···H(n) angle (Δ_c) and τ_c [angle between the $C(3) \cdots H(n)$ vector and the enone plane defined by Me-C(2)-C(3)-Me] [H(n) = H(52) in the 4α -ol, and H(82) in the 4β -ol] are quite similar in the two molecules, 78.5 and 53.2° in the 4 α -ol, and 75.2 and 49.3° in the 4β -ol. In the solid state the molecules have well separated, non-parallel C=C bonds; $C(2) \cdots C(7)$ and $C(3) \cdots C(6)$ intramolecular distances are 4.417(2) and 4.400(2) Å, respectively, for the 4α -ol, and $4\cdot490(3)$ and $4\cdot471(3)$ Å for the 4β -ol, outside the limit of $4 \cdot 1$ Å suggested for successful 2 + 2photocycloadditions (Schmidt, 1971; Scheffer & Dzakpasu, 1978). The $O(1) \cdots H(81)$ intramolecular distance in the 4 α -ol is 2.45 (2) Å, similar to those observed in several tetrahydronaphthoquinone derivatives (Phillips & Trotter, 1977), where photochemical conversion is initiated by H abstraction by carbonyl oxygen (Scheffer & Dzakpasu, 1978); the 4β -ol conformer has no C=O···H interaction geometrically favourable for H abstraction by oxygen, and in fact none of the tetrahydronaphthoquinols studied exhibit this pathway in solution or in the solid state (Appel, Greenhough, Scheffer, Trotter & Walsh, 1980).

The reactivity differences therefore appear to be a consequence of the short β -C···H contacts in the 4α -ol vs the longer distances in the 4β -ol, since the latter has no other primary photochemical reaction pathway available.

The photochemical reactions in solution proceed viaintramolecular 2 + 2 cycloaddition for both molecules, probably due to reaction in solution via the higherenergy conformers C, which better fulfil the geometric

	$4\alpha \text{-ol} \\ (\sigma = 0.002 \text{ Å})$	$\begin{array}{c} 4\beta \text{-ol} \\ (\sigma = \\ 0.003 \text{ Å}) \end{array}$		$4\alpha \cdot ol$ $(\sigma = 0 \cdot 1^{\circ})$	4β -ol ($\sigma =$ 0.2°)		$4\alpha \cdot ol$ $(\sigma = 0 \cdot 1^{\circ})$	4β -ol ($\sigma = 0.2^{\circ}$)
$\begin{array}{c} C(1)-C(2)\\ C(2)-C(3)\\ C(3)-C(4)\\ C(4)-C(4a)\\ C(4a)-C(5)\\ C(5)-C(6)\\ C(5)-C(6)\\ C(6)-C(7)\\ C(7)-C(8)\\ C(8)-C(8a)\\ C(1)-C(8a)\\ C(1)-C(8a) \end{array}$	0.002 Å) 1.479 1.339 1.516 1.535 1.538 1.502 1.333 1.502 1.536 1.527	0-003 Å) 1-464 1-347 1-517 1-532 1-537 1-501 1-330 1-503 1-503 1-522 1-522	C(8a)-C(1)-C(2) C(8a)-C(1)-O(1) C(2)-C(1)-O(1) C(1)-C(2)-C(3) C(1)-C(2)-C(21) C(3)-C(2)-C(21) C(2)-C(3)-C(4) C(2)-C(3)-C(31) C(4)-C(3)-C(31) C(3)-C(4)-C(4a) C(4)-C(4a) C(4)-C(4	(0-1°) 118.8 121.3 119.7 120.6 115.9 123.5 122.5 122.6 114.9 114.0	(0-2°) 119.4 119.4 121.0 120.3 116.3 123.4 122.0 122.2 115.9 114.2	C(7)-C(6)-C(61)C(6)-C(7)-C(8)C(6)-C(7)-C(71)C(8)-C(7)-C(71)C(7)-C(8)-C(8a)C(8)-C(8a)-C(1)C(8)-C(8a)-C(4a)C(4a)-C(8a)-C(81)C(81)-C(8a)-C(8)C(81)-C(8a)-C(1)C(8)-C(8)-C(1)	0.1°) 123.7 121.8 123.8 114.4 115.7 110.6 109.8 113.4 108.7 105.0 109.2	0.2°) 122.8 121.7 124.6 113.7 114.7 106.2 109.6 112.0 108.7 110.3
$\begin{array}{c} C(4a)-C(8a)\\ C(1)-O(1)\\ C(2)-C(21)\\ C(3)-C(31)\\ C(4)-O(4)\\ C(4a)-C(41)\\ C(6)-C(61)\\ C(7)-C(71)\\ C(8a)-C(81) \end{array}$	1.549 1.217 1.510 1.506 1.427 1.533 1.506 1.509 1.545	1.547 1.229 1.507 1.497 1.421 1.536 1.513 1.505 1.531	C(3)-C(4)-O(4) C(4a)-C(4)-O(4) C(4)-C(4a)-C(5) C(4)-C(4a)-C(41) C(4)-C(4a)-C(8a) C(41)-C(4a)-C(5) C(41)-C(4a)-C(8a) C(5)-C(4a)-C(8a) C(5)-C(6)-C(6) C(5)-C(6)-C(7) C(5)-C(6)-C(61)	110.9 106.9 109.4 108.7 109.7 109.1 111.1 108.7 115.1 122.3 114.0	106.8 111.7 109.1 109.7 108.6 109.9 111.1 108.5 116.1 122.3 114.9	C(4a)-C(8a)-C(1)	109-3	109.8

Table 3. Bond distances (Å) and angles (°)

requirements for intramolecular 2 + 2 cycloaddition than do the observed solid-state conformers A and B. The solid-state/solution reactivity differences in both the 4a-ol and the 4 β -ol thus appear to be due to crystal-lattice control of the photorearrangement via control of molecular conformation, as distinct from control in the sense of a least-motion process as found for 2,3,4a β ,6,7 β ,8a β -hexamethyl-4a,7,8,8a-tetrahydro-1,4-naphthoquinone (Appel, Greenhough, Scheffer & Trotter, 1979; Greenhough & Trotter, 1979, 1980), where the completely different unimolecular photorearrangements in solution and the solid state are assumed to proceed from the same molecular conformation.

The bond lengths and angles in both molecules (Table 3) have normal values. The six-membered rings all have half-chair conformations: in the 4α -ol molecule, C(4a) and C(8a) deviate by -0.48 and 0.25Å from the mean plane of C(1) to C(4), and by 0.42 and -0.33 Å from the mean plane of C(5) to C(8); the corresponding displacements in the 4β -ol are -0.53and 0.20, and 0.33 and -0.42 Å. The carbonyl groups are only slightly non-planar, with C(1) out of plane by 0.03 Å (23 σ) in the 4 α -ol, and by 0.02 Å (9 σ) in the 4β -ol. The twist conformation produces two close intramolecular $H \cdots H$ contacts in the 4β -ol, $H(52) \cdots H(O4) = 2 \cdot 18 (4)$ and $H(82) \cdots H(4) =$ 2.27(3) Å; such short contacts are not present in the 4α -ol.

Molecules in both crystals are linked into chains along **a** by $O(4) - H(O4) \cdots O(1)$ intermolecular hydrogen bonds, $O(4) \cdots O(1) = 2.802$ (1) and 2.846 (2) Å in the 4 α -ol and 4 β -ol, respectively, H(O4)...O(1) = 1.98 (2) and 2.06 (3) Å, $O(4)-H(O4)\cdots O(1) = 171$ and 167°. Packing diagrams have been deposited.

Acknowledgement is made to the donors of the Petroleum Research Fund, administered by the American Chemical Society, and to the Natural Sciences and Engineering Research Council Canada for financial support. We thank the University of British Columbia Computing Centre for assistance, and J. R. Scheffer, W. K. Appel and L. Walsh for crystals and photochemical results.

References

- APPEL, W. K., GREENHOUGH, T. J., SCHEFFER, J. R. & TROTTER, J. (1979). J. Am. Chem. Soc. 101, 213-215.
- Appel, W. K., Greenhough, T. J., Scheffer, J. R., TROTTER, J. & WALSH, L. (1980). J. Am. Chem. Soc. 102. 1158-1161.
- BECKER, P. J. & COPPENS, P. (1974). Acta Cryst. A30, 129-147.
- BECKER, P. J. & COPPENS, P. (1975). Acta Cryst. A31, 417-425.
- CROMER, D. T. & MANN, J. B. (1968). Acta Crvst. A24. 321-324.
- GREENHOUGH, T. J. & TROTTER, J. (1979). Acta Cryst. B35, 3084-3087.
- GREENHOUGH, T. J. & TROTTER, J. (1980). Acta Cryst. B36, 368-373.
- LADD, M. F. C. (1978). Private communication.
- MAIN, P., HULL, S. E., LESSINGER, L., GERMAIN, G., DECLERCQ, J. P. & WOOLFSON, M. M. (1978). MULTAN 78. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data. Univs. of York, England, and Louvain, Belgium.
- PHILLIPS, S. E. V. & TROTTER, J. (1976). Acta Cryst. B32, 3088-3091.
- PHILLIPS, S. E. V. & TROTTER, J. (1977). Acta Cryst. B33, 996-1003.
- SCHEFFER, J. R. & DZAKPASU, A. A. (1978). J. Am. Chem. Soc. 100, 2163-2173.
- SCHMIDT, G. M. J. (1971). Pure Appl. Chem. 27, 647-678.
- STEWART, R. F., DAVIDSON, E. R. & SIMPSON, W. T. (1965). J. Chem. Phys. 42, 3175-3187.

Acta Cryst. (1980). B36, 1835-1839

Crystal Structures of 2,3,4a,8a,8a,8-Tetramethyl-4a,5,8,8a-tetrahydro-1-naphthoquin- 4β -ol* and Its Solid-State Photoproduct

BY TREVOR J. GREENHOUGH AND JAMES TROTTER

Department of Chemistry, University of British Columbia, Vancouver, BC, Canada V6T 1W5

(Received 5 October 1979; accepted 10 March 1980)

Abstract

Crystals of the naphthoquinol are monoclinic, $P2_1/n$, a = 7.319(2), b = 11.140(1), c = 15.510(3) Å,

* IUPAC name: 4\beta-hydroxy-2,3,4a\beta,8a\beta-tetramethyl-4a,5,8,8atetrahydro-1(4H)-naphthalenone.

 $\beta = 96.41 (3)^{\circ}$, Z = 4, R = 0.048 for 931 reflexions. The photoproduct is triclinic, $P\overline{1}$, a = 6.981(1), b = 7.619 (2), c = 11.776 (2) Å, $\alpha = 77.70$ (1), $\beta = 85.58 (1), \gamma = 79.54 (2)^{\circ}, Z = 2, R = 0.070$ for 1126 reflexions. The naphthoquinol molecule has a conformation which has the hydroxyl group pseudo-